Scholars Journal of Applied Medical Sciences (SJAMS)

Abbreviated Key Title: Sch. J. App. Med. Sci. ©Scholars Academic and Scientific Publisher A Unit of Scholars Academic and Scientific Society, India www.saspublishers.com ISSN 2320-6691 (Online) ISSN 2347-954X (Print)

Biochemistry

Hemolysis- An Interference or An Influence on Routine Biochemical Parameters Dr. K. R. Minu Meenakshi Devi¹, Dr. M. C. Archana^{2*}, Dr. R. Shanthi , Dr. R. Mahalakshmi, Dr. R. Lalitha, Dr. K. Pramila

¹Assistant Professor, Department of Biochemistry, Stanley Medical College, Chennai, Tamilnadu, India ²Assistant Professor, Department of Biochemistry, Govt. Omandurar Medical College, Chennai, Tamilnadu, India

Original Research Article

*Corresponding author Dr. M. C. Archana

Article History Received: 31.12.2017 Accepted: 06.01.2018 Published: 30.01.2018

DOI: 10.36347/sjams.2018.v06i01.056



Abstract: Hemolysis is one of the most common reason for inadvertent reports, and sample rejection in clinical laboratory. The aim of this study is to know the effect of hemolysis on routine biochemical parameters. Fifteen healthy volunteers are selected for this study. Venous plasma is collected and grouped into four levels of hemolysis based on hemoglobin concentration : Group I: 0-0.10 g/L, Group II: 0.10-0.50 g/L, Group III: 0.51-1.00 g/L, Group IV: 1.01-2.50 g/L, Group V: 2.51-4.50 g/L. Lysis is achieved by mechanical agitation. The analytes are measured in Automated Biochemistry Analyser. Aspartate aminotransferase (AST) levels are significantly affected even at undetectable levels of hemolysis.Variations of potassium and total bilirubin are observed in moderately hemolyzed samples (hemoglobin >1 g/L). Alanine aminotransferase (ALT), Cholesterol, and Inorganic phosphate (Pi) concentrations are not interfered up to severely hemolyzed levels (hemoglobin: 2.5-4.5 g/L). Albumin, Alkaline phosphatase (ALP), Amylase, Chloride, HDL-cholesterol, Creatine kinase (CK), Glucose, Total protein, Triglycerides, and Uric acid levels vary significantly but within allowable limits. Hemolysis causes alteration of many parameters in Biochemistry. It is therefore preferable to do hemoglobin estimation in plasma for analytes with potential interference, to avoid wrong reporting of results. Keywords: Hemolysis, sample rejection, Biochemistry parameters, CLIA, AST.

INTRODUCTION

Prevention of medical errors is a goal of health care. With the advent of automation, the issue of laboratory errors from preanalytical variables has received a great deal of attention.

Among the major preanalytical variables influencing patient results, sample hemolysis exerts a major role. Hemolytic samples are a common occurrence in laboratory practice.

Hemolysis is the most common reason for sample rejection, account for 3.3% of all routine samples and ~ 60% of the rejected samples [1]. Hemolysis can occur in vivo or in vitro. In vivo hemolysis accounts for only 3.2% [2].The incidence of in vitro hemolysis is more and often preventable if due care taken in specimen collection, transportation, & processing [3-5].

Hemolysis can be an interference or an influence factor in the laboratory analysis. Hemolysis causes release of cellular constituents into the plasma which can either falsely elevate the concentration of certain analytes or have a dilutional effect resulting in lower values. Moreover, hemolysis causes spectrophotometric interference of the analytes [6].

Available online at https://saspublishers.com/journal/sjams/home

Conversely, even if the hemolysis is invisible also, there can be discharge of cellular contents into plasma [7]. So invisible hemolysis can cause false results. Some of the analytes are not affected by hemolysis. Ignorance of the above fact can result in inadvertent rejection of samples.

Since the knowledge of possible effects of hemolysis is important for correct interpretation of the results, the aim of the present work is to evaluate the influence of hemolysis on routinely used biochemical tests.

MATERIALS & METHODS

15 healthy volunteers were enrolled in this study. 6ml of venous blood was collected in 5 different heparinised tubes.

Hemolysis was achieved by mechanical trauma. In 4 of the tubes, hemolysis was achieved by pushing forcefully through the needle 3, 6, 9, 12 times respectively to get varying levels of hemolysis. This

method of cell lysis was almost similar to the mechanical disruption of erythrocytes during blood collection. They were all centrifuged at 1000 x g for 10 min. The supernatant plasma removed and was recentrifuged at 1200 x g for 20 min.

The free hemoglobin of the samples were measured colorimetrically by cyanmethemoglobin method [8]. Absorbance was measured at 540nm.

The samples were categorised into five groups, based on free hemoglobin concentrations as: Group I - 0-0.1g/L; no hemolysis (n=10) Group II-0.1-1.0g/L; slight hemolysis (n=10) Group III-1.0-2.5g/L; mild hemolysis (n=14) Group IV-2.5-4.5g/L; moderate hemolysis (n=15) Group V-4.5-6.5g/L; severe hemolysis (n=11)

The exclusion criteria of a hemolyzed sample was the concentration of its free Hb concentration discordant with the degree of hemolysis determined for each group, so the number of samples were not equal at each group. Reference Range for plasma free hemoglobin: 0-0.1g/L.

Plasma concentrations of glucose, urea, creatinine, AST, ALT, ALP, protein, albumin, bilirubin, amylase, sodium, potassium, calcium, chloride, phosphorus, uric acid, cholesterol, triglycerides, HDL were analysed in all groups of hemolysed samples. The measurement of the analytes were done in Olympus AU480 Automated Biochemistry Analyser.

The effects of hemolysis were evaluated according to the total allowable error recommendations of Clinical Laboratory Improvement Amendments (CLIA'88) [9] (Table-1). CLIA' 88 regulations have established fixed limits for assessing the method and laboratory performance for specific regulated analytes. In practice, the total allowable error for a given analytical method must be less than the limits fixed by CLIA for the analyte in question.

Statistics

To compare the concentration of analytes in the hemolysed and the non hemolysed specimen, percentage of bias was calculated by the formula

Cx- concentration of the analyte in hemolysed sample C1-concentration of the analyte in non hemolysed sample

All analysis were done using Graphpad software Version 6.0 for windows XP. Wilcoxon matched pair signed rank test was used. p value < 0.05 was considered statistically significant.

RESULTS

The median free hemoglobin concentration of the groups I, II, III, IV, V were 0.05, 0.65, 1.5, 3.1, 5.5 g/L respectively. At free hemoglobin concentration of 0.2 g/L, hemolysis was visible by the red color of the plasma. The results of this study were shown in table 1 and also in figure-1.

The AST values show significant increase in linearity with free hemoglobin concentration. Among the electrolytes, potassium increased gradually with increase in hemolysis.



Fig-1: Showing comparison of hemoglobin and measured parameters: AST, potassium, Total Bilirubin

Table-1 shows the method, median value of the analytes and free hemoglobin concentrations for each group, percentage relative bias of the analytes compared to the non hemolysed group, desirable bias $\pm,$ CLIA '88 acceptable limits $\pm.$

		Free	Plasma H	emoglobin	1			
Analyte	Method					Desirable	CLIA ±	
		No	0.65	1.5	3.1	5.5	Bias ±	
		lysis						
		4.3	4.2	4.3	4.5	4.7		
Albumin	BCG method						1.3%	10%
(g/L)			(-3%)	(-0.4%)	(+5.8%)	(+8.2%)		
			P=0.36	p=0.88	p=0.04	p=0.03		
		84.5	82	69.5	66	79		
ALP	PNPP		(+3.6%)	(-6%)	(-14%)	(+18%)	6.4%	30%
(U/L)			P=0.78	p=0.38	P=0.26	P=0.57		
		8	11.5	12	12	13		• • • •
ALT	IFCC without		(+3.0%)	(+4.0%)	(+5.0%)	(+18%)	12%	20%
(U/L)	P5P		P=0.52	p=0.12	p=0.24	p=0.04		
		45	46.5	46	47	51	-	2004
Amylase	CNP triose		(-2.0%)	(-5.3%)	(-	(-	7.4%	30%
(u/L)			p=0.04	p=0.03	10.3%)	15.2%)		
			10.7		p=0.03	p=0.03		
	The second secon	15.5	18.5	25.5	35	50		2004
AST	IFCC without		(+30%)	(+60%)	(+116%)	(+201%)	5.4%	20%
(U/L)	P5P	0.47	P=0.04	P=0.009	P=0.003	P=0.002		
D'11 1 1	D:	0.45	0.3	0.43	0.4	0.3		
Bilirubin	Diazo,		(-8%)	(-	(-35%)	(-80%)	100/	2004
total	colorimetric		P=0.02	14.5%)	P=0.02	P=0.019	10%	20%
(mg/dl)		0.0	0.0	P=0.03	0.1	0.1		
Calaina	OCDC	9.2	9.2	9.1 $(+1, 10/)$	9.1	9.1	0.90/	0.25
Calcium Tatal	OCPC method		(-2.0%)	(+1.1%)	(-0.4%)	(+1.5%)	0.8%	0.25mmol/L
10tal			P=0.49	P=0.98	P=0.43	P=0.86		
(mg/dl)		065	065	07	07	00		
Chlorida	Colorimatria	90.3	90.3 (1.8%)	71 (62%)	7/ (1)	77 (26%)	0.5%	504
(mmol/L)	Colorimetric		(-1.070) D=0.84	(-0.270)	(-4.270) P=0.28	(-2.0%)	0.370	570
(IIIIIOI/L)		133	1 -0.04	1_0.40	1 -0.20	1/0		
		133	(1.0%)	(14)	(1550%)	147		
Cholesterol	Enzymatic		P = 0.95	(+4.170) P-0.57	(+3.3%) P-0.32	(+7.0%) P-0.04	4.0%	10%
(mg/dl)	colorimetric		1 -0.75	1-0.57	1-0.52	1 -0.07	T.U/U	10/0

Table-1: Effects	of hemolysis on	various clinical	chemistry	parameters
			•	

	Table-1: Effects of h	nemolysi	s on various	s clinical ch	emistry para	meters (Con	tinuation)	
		Free P	lasma Hem					
Analyte	Method	No lysis	0.65	1.5	3.1	5.5	Desirable Bias ±	CLIA ±
Creatinine	Jaffe's / alkaline picrate	0.7	0.8 (-1.7%) P=0.68	0.8 (0.0%) P=0.24	0.8 (0.0%) P=0.36	0.8 (-2.3%) P=0.83	3.4%	15%
Glucose	GOD	86	89 (-1.0%) P=0.03	79 (-4.2%) P=0.01	85 (-0.9%) P=0.001	82 (-7.8%) P=0.003	2.2%	10%
HDL (mg/dl)	Direct	46	45 (+2.3%) P=0.46	46 (+2.0%) P=0.97	49 (-1.0%) P=0.07	50 (-0.8%) P=0.04	5.2%	30%
Phosphorus (mg/dl)	UV Molybdate	3.5	3.7 (+0.4%) P=0.57	3.7 (+1.6%) P=0.98	3.8 (+3.5%) P=0.04	4.2 (+10.3%) P=0.005	3.2%	10%
#Potassium (mmol/L)	Ion Selective Electrode	3.8	4.3 (+6.7%) P=0.04	4 (+3.1%) P=0.005	4.5 (+22.1%) P=0.003	5.4 (43.4%) P=0.002	1.8%	0.5 mmol/L
Protein Total (g/L)	Biuret	8.4	8.5 (+1.9%) P=0.28	8.4 (+0.7%) P=0.64	8.8 (+3.5%) P=0.08	8.5 (+1.45) P=0.02	1.2%	10%
*Sodium (mmol/L)	Ion Selective Electrode	142	139 (-1.1%) P=0.15	141 (-0.7%) P=0.13	140 (-0.6%) P=0.08	140 (-0.5%) P=0.31	0.3%	4 mmol/L
Triglycerides (mg/L)	Enzymatic, Colorimetric	67	67 (-1.3%) P=0.94	73 (-1.6%) P=0.69	68 (-0.8%) P=0.56	72 (-18.2%) P=0.05	10.7%	25%
Urea (mg/dl)	Urease, UV	22	20 (-6.2%) P=0.57	21 (-1.9%) P=0.48	20 (-1.4%) P=0.43	23 (+3.8%) P=0.82	5.5%	9%
Uric acid (mg/dl)	Uricase, UV	5	4.3 (-14%) P=0.57	5.3 (+1.5%) P=0.62	5.5 (+12%) P=0.37	5 (+1.3%) P=0.03	4.8%	17%

K. R. Minu Meenakshi Devi et al.,	Sch. J. App. Med.	Sci., Jan 2018; 6(1D): 272277
-----------------------------------	-------------------	----------------------	-----------

+To compare the values both CLIA'88 allowable limits and Analytical Quality Specifications Desirable bias [11, 20] were given.

*To compare the values of sodium, limits % was converted to mmol/L:1.1%(1.5 mmol/L),0.7%(0.99 mmol/L),0.6%(0.85 mmol/L),0.5%(0.71 mmol/L).

#To compare the values of potassium, limits % was converted to mmol/L: 6.7% (0.25 mmol/1), 3.1%(0.11mmol/L), 22.1%(0.83mmol/L),43.4%(1.65mmol/L).

^To	compare	the	values	of	calcium,	limits	%	was	converted	tommol/L:
2%(0.18mmol/L),1.1%(0.1mmol/L),0.4%(0.03mmol/L),1.3%(0.11mmol/L).										

Table-1 summarises the effects of hemolysis on various clinical chemistry parameters. As expected the AST, Potassium values increase linearly with increase in free hemoglobin. Total Bilirubin shows a gradual decrease shown in figure-1. amylase, calcium, chloride, HDL-cholesterol, creatinine, glucose, sodium, total protein, triglycerides, urea and uric acid were lower than the CLIA allowable limits even upto 5.5g/L of free haemoglobin, although some differences were statistically significant (p \leq 0.05).

Of the analytes evaluated, the bias recorded for albumin, alkaline phosphatase, alanine transaminase,

Available online at https://saspublishers.com/journal/sjams/home

Value of inorganic phosphorus increase significantly with increase in haemoglobin concentrations.

DISCUSSION

Interference of hemolysis is due to the release of hemoglobin and other cellular constituents that may falsely increase many of the analyte concentrations because of large differences between intracellular and extracellular concentrations. In this study, the concentrations of AST, potassium, phosphorus showed significant increase as expected from the previous studies [10-13].

Free hemoglobin with its pseudo-peroxidase activity interferes in the bilirubin procedure by inhibiting the diazonium color formation. In our study, we found statistically low values for all groups as in literature [14].

Some of the analytes have a dilutional effect due to discharge of cellular contents showing a false decrease as is the case of glucose, sodium and calcium in this study [19]. In addition to the dilutional effect, the decrease in glucose can also be attributed to the premature decomposition of hydrogen peroxide by hemglobin.

Chemical interference of free hemoglobin occurs in a variety of analytic reactions and methods and analyte concentration dependent spectrophotometric interference [15], due to an increase of the optical absorbance or a change in the blank value, especially for laboratory tests employing measurements at 415, 540 and 570 nm, where hemoglobin absorbs more strongly [16, 17].

The first step to eliminate interference is recognising its existence. Visually, hemolysis is defined as free hemoglobin concentrations above 0.2 g/L [18], which confers a detectable pink to red hue to serum or plasma. But AST concentrations are affected even before hemolysis is visible. Nowadays serum hemolysis index is a popular solution for interference detection preanalytically. Manufacturers give the list of test-specific serum indices for hemolysis, lipemia and bilirubin interferences.

CONCLUSION

The clinical usefulness of laboratory test results depends on accuracy and precision. The presence of endogenous or exogenous substances in body fluids can adversely affect the determination of many analytes in laboratory practice. We conclude that hemolysis affects many of the laboratory parameters most notably AST, potassium, phosphorus and total bilirubin. For other analytes; albumin, ALP, amylase, chloride, HDL-cholesterol, glucose, total protein, triglycerides, and uric acid, differences were statistically significant, but remained within the CLIA limits.

We therefore recommend routine estimation of free hemoglobin for those analytes which have a significant interference.

REFERENCES

- Lippi G, Blanckaert N, Bonini P, Green S, Kitchen S, Palicka V, Vassault AJ, Plebani M. Haemolysis: an overview of the leading cause of unsuitable specimens in clinical laboratories. Clinical Chemistry and Laboratory Medicine. 2008 Jun 1;46(6):764-72.
- Koseoglu M, Hur A, Atay A, Cuhadar S. Effects of hemolysis interference on routine biochemistry parameters. Biochemia medica: Biochemia medica. 2011 Feb 15;21(1):79-85.
- 3. Haemolysis: The No.1 Reason for Specimen Rejection; linkout resources.
- 4. Lemery L. Oh, no! it's hemolyzed!'what, why, who, how. What. Why. Who. How. 1998:24-5.
- 5. Stankovic AK, Smith S. Elevated serum potassium values: the role of preanalytic variables. Pathology Patterns Reviews. 2004 May 1;121(suppl_1):S105-12.
- 6. Carraro P, Servidio G, Plebani M. Hemolyzed specimens: a reason for rejection or a clinical challenge?. Clinical chemistry. 2000 Feb 1;46(2):306-7.
- 7. Thomas L. Haemolysis as influence and interference factor. eJIFCC vol 13 no 4.
- 8. Burtis CA, Ashwood ER, Bruns DE. Tietz textbook of clinical chemistry and molecular diagnostics-ebook. Elsevier Health Sciences; 2012 Oct 14.
- Peddecord KM, Hammond HC. Clinical laboratory regulation under the Clinical Laboratory Improvement Amendments of 1988: can it be done?. Clinical chemistry. 1990 Dec 1;36(12):2027-35.
- 10. Effects of Hemolysis on Clinical Specimens;www.calgary labservices.com.
- 11. Lippi G, Salvagno GL, Montagnana M, Brocco G, Guidi GC. Influence of hemolysis on routine clinical chemistry testing. Clinical Chemistry and Laboratory Medicine (CCLM). 2006 Mar 1;44(3):311-6.
- Yücel D, Dalva K. Effect of in vitro hemolysis on 25 common biochemical tests. Clinical chemistry. 1992 Apr 1;38(4):575-7.
- 13. Grafmeyer D, Bondon M, Manchon M, Levillain P. The influence of bilirubin, haemolysis and turbidity on 20 analytical tests performed on automatic analysers. Clinical Chemistry and Laboratory Medicine. 1995;33(1):31-52.
- Sonntag O. Haemolysis äs an Interference Factor in Clinical Chemistry. J. Clin. Chem. Clin. Biochem. 1986;24(2):127-39.
- 15. Guder WG. Haemolysis as an influence and interference factor in clinical chemistry. Journal of

clinical chemistry and clinical biochemistry. Zeitschrift fur klinische Chemie und klinische Biochemie. 1986 Feb;24(2):125.

- Kroll MH, Elin RJ. Interference with clinical laboratory analyses. Clinical chemistry. 1994 Nov 1;40(11):1996-2005.
- Steen G, Vermeer HJ, Naus AJ, Goevaerts B, Schoenmakers CH. Multicenter evaluation of the interference of hemoglobin, bilirubin and lipids on Synchron LX-20 assays. Clinical Chemistry and Laboratory Medicine (CCLM). 2006 Apr 1;44(4):413-9.
- Tietz NW. Specimen collection and processing; sources of biological variation. Textbook of Clinical Chemistry, 2nd Edition, WB Saunders, Philadelphia, PA. 1994;4:416-7.
- 19. Frank JJ, Bermes EW, Bickel MJ, Watkins BF. Effect of in vitro hemolysis on chemical values for serum. Clin Chem 1978;24:1966-70.
- Ricós C, Alvarez V, Cava F, Garcia-Lario JV, Hernandez A, Jimenez CV, Minchinela J, Perich C, Simon M. Current databases on biological variation: pros, cons and progress. Scandinavian journal of clinical and laboratory investigation. 1999 Jan 1;59(7):491-500.