

Research Article

## Empirical Study on the Main Physical Items of Nano Oil in an Isolated Vertical Pipe

Mohammad Yaz<sup>1</sup>, Farshad Farahbod<sup>2\*</sup><sup>1</sup>Department of Chemical Engineering, Shahreza Branch, Islamic Azad University, Shahreza, Iran<sup>2</sup>Department of Chemical Engineering, Firoozabad Branch, Islamic Azad University, Firoozabad, Iran

### \*Corresponding author

Farshad Farahbod

Email: [mf\\_fche@yahoo.com](mailto:mf_fche@yahoo.com)

**Abstract:** Finding the properties of crude oil after addition of nano particle is considered in this work. Results are shown in Figures indicate on the positive effect of nano particles in the oil sweetening processes. The amount of H<sub>2</sub>S removal increases about 0.19 and 0.11 for both oil types of heavy and light, respectively. Temperature varies in range of 80 C to 120 C and pressure is adjusted between 50 atm to 250 atm. Addition of nano particle shows the decrease in the amount of viscosity about 70% and 50% for light and heavy oil, respectively. However, nano particle in oil doesn't improve the recovery percentage and also asphalt precipitation. Due to the results, the optimums of properties can be evaluated. Ultimately, the experimental results show the effect of addition of nano particle in the percentage of oil recovery versus the input energy for sample A and sample B, respectively.

**Keywords:** Nano; Petroleum; Hydrogen sulfide; Light; Heavy; Properties.

### INTRODUCTION

Crude oil, liquid petroleum that is found accumulated in various porous rock formations in Earth's crust and is extracted for burning as fuel or for processing into chemical products. Although it is often called "black gold," crude oil has ranging viscosity and can vary in color to various shades of black and yellow depending on its hydrocarbon composition.

In 1847, the process to distill kerosene from crude oil was invented by James Young [1]. He noticed a natural petroleum seepage in the Riddings colliery at Alfreton, Derbyshire from which he distilled a light thin oil suitable for use as lamp oil, at the same time obtaining a thicker oil suitable for lubricating machinery. In 1848 Young set up a small business refining the crude oil [2]. Young eventually succeeded, by distilling cannel coal at a low heat, in creating fluid resembling petroleum, which when treated in the same way as the seep oil gave similar products. Young found that by slow distillation he could obtain a number of useful liquids from it, one of which he named "paraffine oil" because at low temperatures it congealed into a substance resembling paraffin wax.

Another early refinery was built by Ignacy Łukasiewicz, providing a cheaper alternative to whale

oil. The demand for petroleum as a fuel for lighting in North America and around the world quickly grew [3].

Desulfurization of crude oil is an important process used in a 16 petroleum refinery to reduce the sulfur concentration and production of fuel products such as gasoline, jet fuel, kerosene, diesel, 18 and heating oil [1]. So, the resulting fuels meet environmental 19 protection standards [2]. Sulfur and its derivatives existed in 20 extracted petroleum are released in air during burning and heating 21 process and disturb the laborer's health [3]. The challenge of 22 fulfilling the world's growing transportation energy needs is no 23 longer a simple issue of producing enough liquid hydrocarbon 24 fuels [4]. This challenge is instead accentuated by a complex 25 interplay of environmental and operational issues. Severe changes 26 in the climate, which lead to disasters, are caused by the pollutants 27 from fuel burning [5]. Environmental issues include societal 28 demands that liquid hydrocarbon fuels are clean and less polluting 29 [6,7]. The emergence of new refining processes and the increasing 30 use of new forms of energy production, e.g., fuel cells, exemplify 31 operational issues [7,8]. Although studies on new forms of renew- 32 able energies have been investigated recently [7,8], fossil fuels are 33 the major source of energy yet. Together,

these trends are driving the need for deep desulfurization of diesel and jet fuels. 1.1 Desulfurization Processes. In the past two decades, petroleum refining has changed extensively and the fortunes of hydro treating, in particular, have witnessed a sea change [8,9]. Hydro-treaters now occupy a central role in modern refineries and more than 50% of all refinery streams now pass through hydro-treaters for conversion, finishing, and pretreatment purposes [10,11]. Hydro-desulfurization is the largest application of catalytic technology in terms of the volume of material processed. On the basis of usage volume, Hydrodesulfurization catalysts are ranked third behind catalysts used for automobile emission control and fluid catalytic cracking. Commercial hydro-treating catalysts are, typically, Zn, Co, or Mo. For example, Mo, known for its high hydrogenation activities, is preferred as a promoter when feed stocks containing high amounts of nitrogen and aromatics need to be processed. Catalyst based on Cr (III) shows high selectivity for sulfur removal in gasoline treatment [12]. It seems nano-particles such as metal oxides can promote the heating and cooling process [12–14]. Nano metal oxide application in heating and cooling is extended in various fields related to the exploring, storage, refining, and recovery of oil [15]. For example, the nano substances like metal oxides can enhance the thermal stability of some of the materials [16]. Today, about 90% of vehicular fuel needs are met by oil. Petroleum also makes up 40% of total energy consumption in the U.S., but is responsible for only 1% of electricity generation. Petroleum's worth as a portable, dense energy source powering the vast majority of vehicles and as the base of many industrial chemicals makes it one of the world's most important commodities. The viability of the oil commodity is controlled by several key parameters, number of vehicles in the world competing for fuel, the quantity of oil exported to the world market (export land model), Net Energy Gain (economically useful energy provided minus energy consumed), political stability of oil exporting nations, and ability to defend oil supply lines. The top three oil-producing countries are Russia, Saudi Arabia, and the U.S. About 80% of the world's readily accessible reserves are located in the Middle East, with 62.5% coming from the Arab 5: Saudi Arabia, UAE, Iraq, Qatar and Kuwait. A large portion of the world's total oil exists as unconventional sources, such as bitumen in Canada and extra heavy oil in Venezuela. While significant volumes of oil are extracted from oil sands, particularly in Canada, logistical and technical hurdles remain, as oil extraction requires large amounts of heat and water, making its net energy content quite low relative to conventional crude oil. Thus, Canada's oil sands are not expected to provide more than a few million barrels per day in

the foreseeable future. In its strictest sense, petroleum includes only crude oil, but in common usage it includes all liquid, gaseous, and solid hydrocarbons. An oil well produces predominantly crude oil, with some natural gas dissolved in it. Because the pressure is lower at the surface than underground, some of the gas will come out of solution and be recovered (or burned) as associated gas or solution gas. At surface conditions, these will condense out of the gas to form natural gas condensate, often shortened to condensate. Condensate resembles petrol in appearance and is similar in composition to some volatile light crude oils. The proportion of light hydrocarbons in the petroleum mixture varies greatly among different oil fields, ranging from as much as 97% by weight in the lighter oils to as little as 50% in the heavier oils and bitumens. The exact molecular composition varies widely from formation to formation but the proportion of chemical elements varies over fairly narrow limits as follows. Distillation, the process by which oil is heated and separated into different components, is the first stage in refining. Crude oil is any naturally occurring flammable mixture of hydrocarbons found in geologic formations, such as rock strata. Most petroleum is a fossil fuel, formed from the action of intense pressure and heat on buried dead zooplankton and algae. Technically, the term petroleum only refers to crude oil, but sometimes it is applied to describe any solid, liquid, or gaseous hydrocarbons. Petroleum consists primarily of paraffins and naphthenes, with a smaller amount of aromatics and asphaltenes. The exact chemical composition is a sort of fingerprint for the source of the petroleum [17, 18]. Petroleum assay need to be focused in selection of the treatment process and sulfur removal from the extracted oil [17, 18].

Access to oil was and still is a major factor in several military conflicts of the twentieth century, including World War II, during which oil facilities were a major strategic asset and were extensively bombed [19].

The recovery percentage of nano crude oil and also, the recovery percentage of simple crude oil are investigated in this paper. The recovery percentage of nano and simple crude oil in different amounts of heat values are measured in this paper.

## MATERIALS AND METHODS

Two samples of crude oil are used in this experimental work. Recovery of heavy oil from sand beds is investigated in this work, experimentally. Same sand beds are provided and set in the experimental vessel one by one for each light oil and heavy oil with and without nano particles, separately. Sand bed is weighted before experiment. Oil stream passed through the bed at the same conditions. Sand bed is weighted

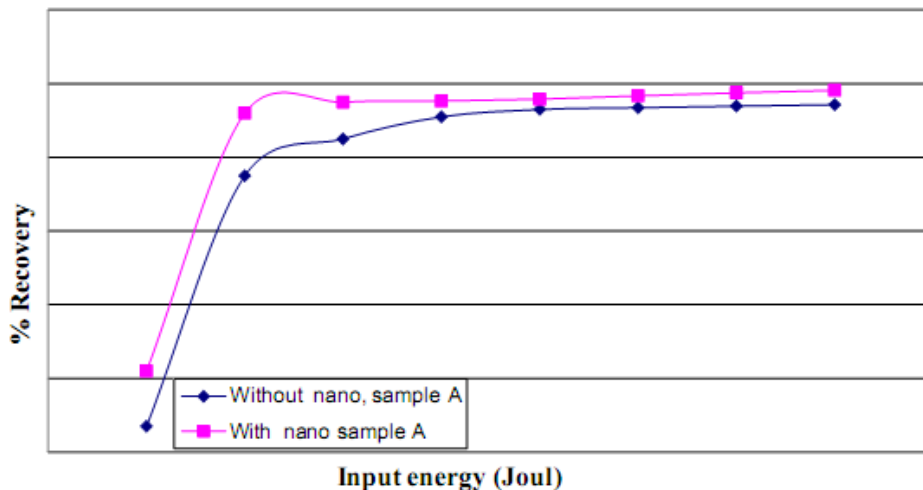
after passing the oil. The vessel, is heated till the amounts of oil recovery are leveled out at different temperatures.

**RESULTS AND DISCUSSION**

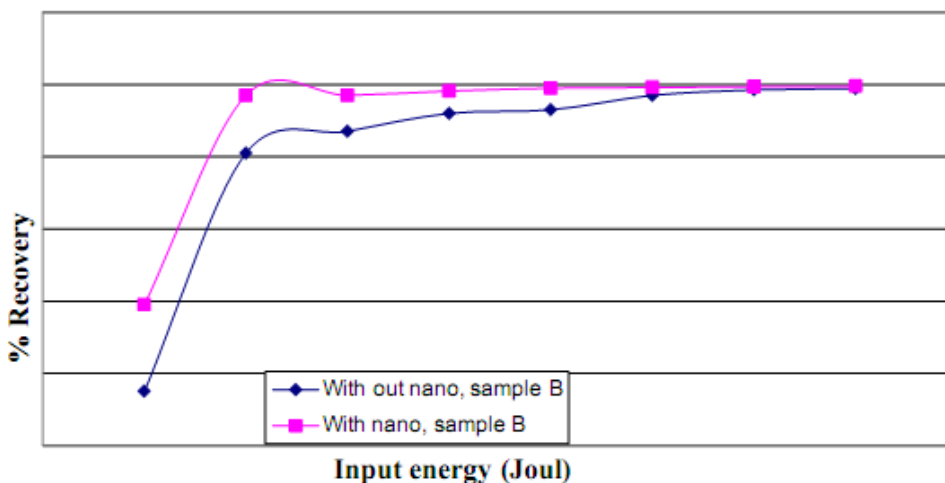
Properties of crude oil are important in refinement processes in refineries. Investigations have been focused on separation methods which lead to the refined oil with acceptable quality. Addition of nano aluminum particles in two kinds of medium and light

oil samples is investigated and viscosity and recovery% of oils are surveyed. Also, H2S removal and asphaltene precipitation in different temperatures and pressures are investigated, experimentally.

The Figures 1 and 2 show the effect of addition of nano particle in the percentage of oil recovery versus the input energy for sample A and sample B, respectively.



**Fig-1: Recovery percentage of sample A versus input energy**



**Fig-2: Recovery percentage of sample B versus input energy**

Operating conditions such as temperature and pressure are effective on the percentage of asphaltene precipitation.

Figure 3 shows the effect of nanoparticles in variations of oil viscosity for sample A, with various

temperatures. Results show the increase in temperature from 80 C to 120 C decreases the amount of viscosity. Nano particle decreases the viscosity about 50%, averagely.

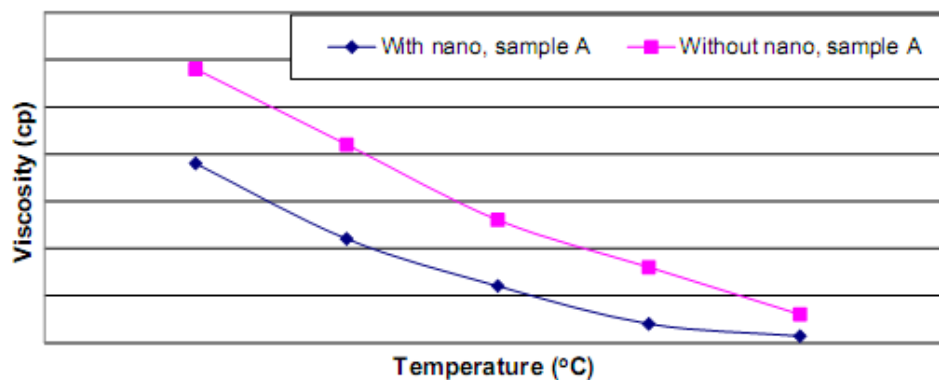


Fig-3: Viscosity versus temperature of sample A with and without nano particle

Figure 4 shows the variations in viscosity of oil type B with temperature with and without nano particles. Increasing temperature from 80 C to 120 C decreases the viscosity from 22 cp to 1.3 cp and 12 cp

to 0.13 cp for sample B without nano and with nano, respectively. Nano particle decreases the viscosity of oil sample B about 70%, averagely comparing with the viscosity of oil sample B without nano.

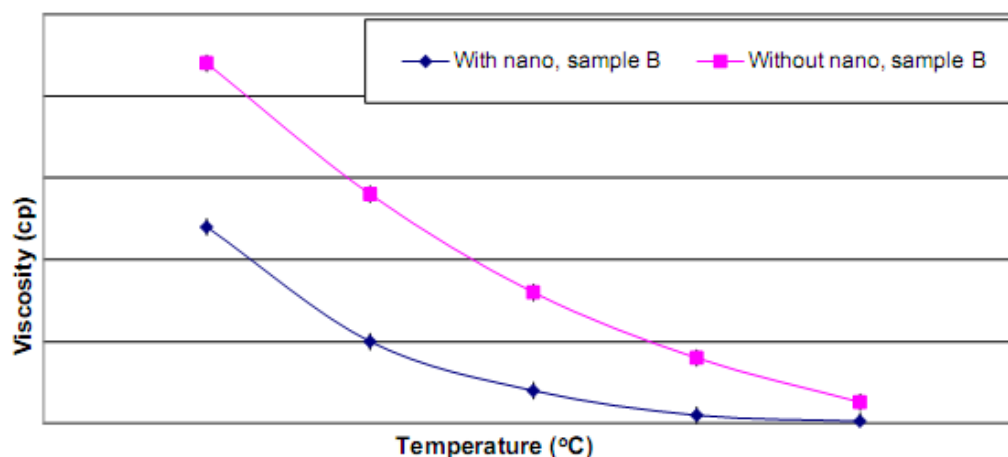


Fig- 4: Viscosity versus temperature of sample B with and without nano particle

The constant temperature for oil means the constant kinetic energy for one type of oil with and without nano particle. So, nano particle seems to affect the cohesion and adhesion forces in oil contains nano particle to change the viscosity, according to the Figures 3 and 4.

### CONCLUSIONS

Application of nano particles to introduce the novel oil refinement techniques is studied, in this work. The effect of nano aluminum oxide particle on the important factors in oil refinery process for light oil (API= 32.95) and heavy oil (API =21.45) is considered. Experiments are held to investigate the viscosity, the amount of asphaltene precipitation, oil recovery from the sand bed and H<sub>2</sub>S removal percentage for oil samples containing nano particles. Also, the values of properties are obtained at different temperatures and pressures. Results show increasing temperature from 80 C to 120 C decreases the viscosity from 22 cp to 1.3 cp and 12 cp to 0.13 cp for light crude oil without nano and with nano, respectively. Nano particle decreases the viscosity

of light crude oil about 70%, averagely comparing with the viscosity of light crude oil without nano.

### REFERENCES

1. Shadi WH, Mamdouh TG, Nabil E. Heavy crude oil viscosity reduction and rheology for pipeline transportation. *Fuel* 2010; 89: 1095–100.
2. Martnez-Palou R, Mosqueira ML, Zapata-Rendón B, Mar-Juárez E, Bernal-Huicochea C, Clavel-López JC. Transportation of heavy and extra-heavy crude oil by pipeline: a review. *J Pet Sci Eng.* 2011; 75: 274–82.
3. Weissman JG. Review of processes for downhole catalytic upgrading of heavy crude oil. *Fuel Proc Technol.* 1997; 50: 199–213.
4. Rana MS, Suman V, Ancheyta J, Diaz JAI. A review of recent advances on process technologies for upgrading of heavy oils and residua. *Fuel.* 2007; 86: 1216–31.
5. Naseri A, Nikazar M, Mousavi DSA. A correlation approach for prediction of crude oil viscosities. *J Pet Sci Eng* 2005; 47: 163–74.

6. Hossain MS, Sarica C, Zhang HQ. Assessment and development of heavy-oil viscosity correlations. In: SPE International Thermal Operations and Heavy Oil Symposium, Kalgary. 1–3 November 2005. p. 1–9.
7. Alomair O, Elsharkawy A, Alkandari H. Viscosity predictions of Kuwaiti heavy crudes at elevated temperatures. In: SPE Heavy Oil Conference and Exhibition, Kuwait. 12–14 December 2011. p. 1–18.
8. Barrufet MA, Setiadarma A. Reliable heavy oil-solvent viscosity mixing rules for viscosities up to 450 K, oil-solvent viscosity ratios up to 4 \_ 105, and any solvent proportion. *Fluid Phase Equilib.* 2003; 213: 65–79.
9. Fazaeli R, Aliyan H, Moghadam M, Masoudinia M. “Nano-Rod Catalysts: Building MOF Bottles (MIL-101 Family as Heterogeneous 478 Single-Site Catalysts) Around Vanadium Oxide Ships. *J. Mol. Catal. A: 479 Chem.* 2013;374–375, pp. 46–52. 480.
10. Novochimskii II, Song CH, Ma X, Liu X, Shore L, Lampert J, Farrauto RJ. “Low Temperature H 481 2S Removal From Steam Containing Gas Mixtures With ZnO for Fuel Cell Application—1: ZnO Particles and 482 Extrudates. *Energy Fuel.* 2004;18(2): 576–591. 483.
11. Thamir KI, Rahman MM. “Optimum Performance Improvements of the Combined Cycle Based on an Intercooler-Reheated Gas Turbine. 484 *ASME J. Energy Resour. Technol.* 2015; 137(6):061601. 485.
12. Bera A, Babadagli T. “Status of Electromagnetic Heating for Enhanced Heavy Oil/Bitumen Recovery and Future Prospects: A Review. 486 *Appl. Energy.* 2015; 151: 206–226. 487.
13. Farahbod F, Farahmand S. “Experimental and Theoretical Study of Fluidized Bed for SO 488 2 Recovery as Sulfur From Effluent Gases From Sulfur Production Unit. *Fuel J. Fuel.* 2015; 156: 103–109. 489.
14. Wojtanowicz Andrew K. Special Issue: Recent Studies of Petroleum Wells and Reservoirs,” *ASME J. Energy Resour. Technol.* 2014;136(4):490 p. 040301. 491.
15. Emiliano P, Giuseppe G. “Experimental Determination of Lique- fied Petroleum Gas–Gasoline Mixtures Knock Resistance. *ASME J. Eng. Gas 492 Turbines Power.* 2014;136(12):121502. 493.
16. Heming W, Xianshang L, Lijun Z, Yulu Z, Daohong X. Preparation and Performance of a Fixed Bed Catalyst for the Oxidation 494 of Sodium Mercaptides. *Bull. Chem. React. Eng. Catal.* 2014;9(2): 495 87–92. 496.
17. Hakimi A, Farahbod F. “Experimental Evaluation of Dimensionless Groups for Scale Up of Sulfur Removal Process. *Int. J. Chem. Biomol. 497 Sci.* 2016;2(1): 43–46. 498.
18. Ganguly SK, Das G, Kumar G, Kumar S, Sain B, Garg MO. “Catalytic Oxidation of Mercaptans in Light Oil Sweetening: Kinetics 499 and Reactor Design. *Chem. Eng. Trans.* 2013;32: 661–666. 500.
19. Riazi MR. *Characterization and Properties of Petroleum Fractions*, 1st, ed., American Society for Testing and Materials, West Conshohocken, PA. 2005;501: 254–257.