

Measurement of Radon Exhalation Rates from Soil Samples of Some Villages of Ambala, Haryana

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Abstract: The world is naturally radioactive as some of the materials in the nature are radioactive. Uranium is one of them and can be found in trace amount in most rocks and soils. Uranium decays to radium that leads to the decay of radon. Because radon is a noble gas, a large portion of it is free to migrate away from radium. Radon and its progeny are major contributors in the radiation dose received by general population of the world. A sufficient quantity of radon comes out of rock and soil in the environment. The primary sources of indoor radon are soils and rocks source emanations, emanation from building materials, and entry of radon into a structure from outdoor air. Keeping this in mind the study of radon exhalation rate from some soil samples of the villages of Ambala, Haryana has been carried out using Can Technique. The equilibrium radon concentration in various soil samples varied from 30.82 Bqm⁻³ to 84.52 Bqm⁻³ with an average of 49.02 ± 9.21 Bqm⁻³. The radon mass exhalation rates from the soil samples varied from 1.11 to 3.04 mBq/kg/h with an average of 1.76 ± 0.33 mBq/kg/h and radon surface exhalation rates varied from 25.09 to 68.83 mBq/m²/h with an average of 39.92 ± 7.50 mBq/m²/h. The radon mass and surface exhalation rates of the soil samples of villages of Ambala, Haryana were lower than that of the world wide average.

Keywords: Radon; SSNTD; Mass exhalation rate; Surface exhalation rate.

INTRODUCTION

Radioactivity in the environment is the biggest concern for the human beings. Natural radioactivity is common in the rocks, soil, water, oceans that make up our planet, and in our building materials because of the presence of radioactive nuclides like uranium, thorium, ⁴⁰K and ¹⁴C etc. Since the distribution of these radionuclides is not uniform in our nature so the knowledge of their distribution and radiation levels due to them in the environment is important for assessing the effects of radiation exposure to human beings. Radionuclides such as ²²²Rn and ²²⁰Rn and their daughter products represent sources of internal radiation exposure to man from inhalation from indoor air. The inhalation of short-lived decay products of radon (²²²Rn) accounts on average about 55% of the effective equivalent dose on the human being. ²²²Rn is part of the ²³⁸U decay chain and is frequently generated close to it. Due to the serious public health implications of exposure to high levels of radon, the measurement of radon exhalation rates from some soil samples of villages of Ambala, Haryana has been carried out using canister technique. Our earlier work in different parts of Punjab and Haryana shows that the values of indoor radon, thoron and their progeny and radon exhalation rates are within the safe limits [1-8].

MATERIALS AND METHODS

For the measurement of radon concentration and exhalation rates from soil samples the “Canister Technique” [9] used by many groups [2-3, 6, 10-12] has been adapted. The details of the technique are given elsewhere [6]. The can is tightly closed from the top and sealed as shown in Fig. 1. At the end of the exposure time (~100 days), the detectors are removed and subjected to a chemical etching process in 2.5 N NaOH solution at 60 ± 0.5 °C for 90 minutes in a constant temperature water bath. The etched detectors are thoroughly washed and then immediately after the completion of washing, the red sensitive layer is stripped for the spark counting.

From the track density (track/cm²/day), the radon activity was obtained in Bqm⁻³, using the calibration factor of 0.056 tr. cm⁻² d⁻¹/Bqm⁻³ obtained from an earlier calibration experiment [13]. The radon mass and surface exhalation rates from the soil sample can be calculated by following relation.

$$E_A = \frac{CV\lambda}{A(T + \frac{1}{\lambda}(e^{-\lambda T} - 1))} \quad (1)$$

$$E_M = \frac{CV\lambda}{M(T + \frac{1}{\lambda}(e^{-\lambda T} - 1))} \quad (2)$$

Where C is equilibrium radon activity inside the canister, V and A are volume and area of cross-section of the canister, M is the mass of the sample and λ is the radon decay constant, T is the time of exposure.

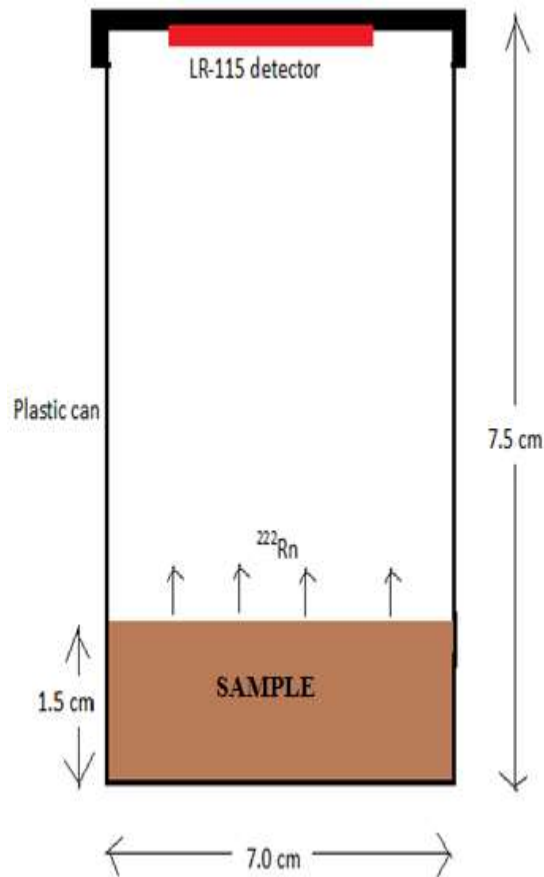


Fig. 1: The plastic can used for the measurement of equilibrium radon concentration and exhalation rates from different soil samples

RESULTS AND DISCUSSION

The values of exhalation rates found were tabulated in Table 1. The radon concentration in various soil samples of Ambala, Haryana varied from 30.82 Bqm⁻³ to 84.52 Bqm⁻³ with an average of 49.02 ± 9.21 Bqm⁻³. The radon mass exhalation rates from the soil samples varied from 1.11 to 3.04 mBq/kg/h with an average of 1.76 ± 0.33 mBq/kg/h and radon surface exhalation rates varied from 25.09 to 68.83 mBq/m²/h with an average of 39.92 ± 7.50 mBq/m²/h.

Table 1: Equilibrium radon concentration, radon mass and surface exhalation rates from soil samples of Ambala, Haryana

Sr. No.	Location	Equilibrium Radon conc. (Bqm ⁻³)	Mass exhalation rate (mBqkg ⁻¹ h ⁻¹)	Surface exhalation rate (mBqm ⁻² h ⁻¹)
1	Vill. Mullana	43.54	1.57	35.46
2	Vill. Mithapur	84.52	3.04	68.83
3	Vill. Saha	44.76	1.61	36.45
4	Vill. Kalpi	30.82	1.11	25.09
5	Vill. Barara	41.46	1.49	33.76
AM ± SE		49.02±9.21	1.76 ± 0.33	39.92 ±7.50

AM(arithmetic mean); SE (standard error) = σ/\sqrt{N} , Where σ is SD (standard deviation) and N is the no. of observations

CONCLUSION

The radon exhalation rates of soil sample collected from the study area are nearly the same having not much difference. The radon exhalation rates from soil samples are less than the world wide average [14]. The measurements indicate normal levels of natural radioactivity in soil samples of the study area.

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